

## Optimization and characterization of solvent-tolerant lipases from *Geobacillus stearothermophilus* CHI1

Optimización y caracterización de lipasas tolerantes a solventes de *Geobacillus stearothermophilus* CHI1

César García Hernández<sup>1</sup>, Cristina Ventura Canseco<sup>1</sup>, Miguel Abud Archila<sup>1</sup>, Sandy Luz Ovando Chacón<sup>1</sup>, Peggy Elizabeth Alvarez Gutiérrez<sup>1,2\*</sup>

<sup>1</sup> Instituto Tecnológico de Tuxtla Gutiérrez/Tecnológico Nacional de México. C.P. 29050, Tuxtla Gutiérrez Chiapas, México.

<sup>2</sup>\*CONAHCYT/ Instituto Tecnológico de Tuxtla Gutiérrez/Tecnológico Nacional de México. C.P. 29050, Tuxtla Gutiérrez Chiapas, México.

[peggy.alvarez@hotmail.com](mailto:peggy.alvarez@hotmail.com)

\*Autor de correspondencia

### Abstract

Lipases are enzymes attractive for their use in enzymatic biotechnology, mainly those produced by extremophiles. Recently, the isolation of thermophilic bacteria *Geobacillus* spp. from the geothermal waters of the crater lake of the "El Chichón" volcano in Mexico was reported. In this work, the lipolytic activity of the *G. stearothermophilus* CHI1 strain was characterized and optimized. This strain was able to produce enzymes with maximum lipolytic activity at 60 °C and 80 °C, with pH values of 5, 9, and 11; showed tolerance of solvents; was able to perform catalysis independently of metal ions; and showed greater affinity towards medium-chain substrates. Results of biochemical characterization and optimization suggest the presence of more than one type of lipolytic activity in *Geobacillus stearothermophilus* CHI1. All these characteristics make these enzymes attractive in biotechnological processes at high temperatures and alkaline pH (e.g., detergent additives), as well as in the understanding of their biological utility in *Geobacillus*.

**Keywords:** *Geobacillus*; extremophiles; lipases; enzyme characterization.

### Resumen

Las lipasas son enzimas atractivas para su uso en biotecnología, principalmente aquellas producidas por extremófilos. Recientemente fueron aisladas bacterias termófilas *Geobacillus* en aguas geotermales del lago del cráter del volcán "El Chichón", en México. En este trabajo se caracterizó y optimizó la actividad lipolítica de la cepa *Geobacillus stearothermophilus* CHI1. Esta cepa fue capaz de producir enzimas con máxima actividad lipolítica a 60 °C y 80 °C, con valores de pH de 5, 9 y 11, además de que demostró ser tolerante a solventes y capaz de realizar catálisis independientemente de iones metálicos. Adicionalmente, mostró mayor afinidad hacia sustratos de cadena media. Los resultados de la caracterización y optimización bioquímica sugieren la presencia de más de un tipo de actividad lipolítica presente en *Geobacillus stearothermophilus* CHI1. Todas estas características hacen que estas enzimas sean atractivas en procesos biotecnológicos a altas temperaturas y pH alcalino (por ejemplo, aditivos detergentes), así como para ayudar a comprender su utilidad biológica en *Geobacillus*.

**Palabras clave:** *Geobacillus*; extremófilos; lipasas; caracterización enzimática.

Recibido: 04 de marzo de 2023

Aceptado: 03 de mayo de 2024

Publicado: 24 de julio de 2024

**Cómo citar:** García Hernández, C., Ventura Canseco, C., Abud Archila, M., Ovando Chacón, S. L., & Alvarez Gutiérrez, P. E. (2024). Optimization and characterization of solvent-tolerant lipases from *Geobacillus stearothermophilus* CHI1. *Acta Universitaria* 34, e4153 doi: <http://doi.org/10.15174/au.2024.4153>

## Introduction

Lipases (E.C. 3.1.1.3) catalyze the hydrolysis of ester bonds in long-chain triglyceride molecules by releasing fatty acids, diglycerides, monoglycerides, and glycerol (Godoy *et al.*, 2022). Lipases are hydrolytic enzymes that catalyze the breakdown of ester bonds of water-insoluble substrates. Lipases are versatile enzymes that undergo chemical reactions such as esterification, transesterification, acidolysis, and aminolysis (Mahfoudhi *et al.*, 2022). The active site of these enzymes generally has a catalytic triad (Lim *et al.*, 2022). There is currently great interest in lipases because of their potential use in numerous industrial applications, including biodiesel production, food flavoring, the detergent industry, cosmetics production, the paper industry, the pharmaceutical industry, and biosensors, among others (Pohanka 2019; Vivek *et al.*, 2022). They are currently the third largest enzyme group in terms of total enzyme market value, after protease enzymes and carbohydrases (Godoy *et al.*, 2022). Many approaches have been used to get lipases with biotechnological applications, from computational analysis (Haman *et al.*, 2021) to recombinant enzymes (Oliart-Ros *et al.*, 2021) and synthetic biology (Gamboa-Melendez *et al.*, 2018).

Lipases are produced by all organisms; however, the most studied and used in industry and biotechnology are lipases obtained from microbial sources (Adetunji & Olaniran, 2021). The biotechnological importance of microbial lipases derives from their greater range of operating conditions (pH, temperature, etc.), and they can be synthesized in less time. Furthermore, their use can be a good alternative for reducing costs in the production of lipases. Optimal growth conditions, however, must first be examined to obtain the maximum yield of microbial lipases (Barik *et al.*, 2022; Godoy *et al.*, 2022). In microorganisms, the production of lipases is subject to the composition of the culture medium and the fermentation conditions, such as the culture temperature, the culture pH, and the speed of agitation, among other factors (Behera *et al.*, 2019). To maximize the production of lipases, statistical tools must be used to find the best culture conditions. Then, statistically based optimization methods, such as response surface design, allow determining which are the most important factors in enzyme production and simultaneously determine optimal response values (Behera *et al.*, 2019; Rmili *et al.*, 2022).

Lipases from extremophilic organisms (called extremozymes) are more useful in some biotechnological applications because of their greater stability under conditions of high temperatures, high salt concentrations, and extreme pH values (Adrio & Demain, 2014). Examples of these enzymes are thermos-alkaline (TA) lipase from *Bacillus thermoleovorans* ID-1 with optimal activity at 70 °C and TA lipase from *Bacillus* sp.; LBN2 with optimal activity at 60 °C and pH 10; TA lipase from *Geotrichum candidum* with enzymatic activity in a pH ranging from 5 to 12; and TA lipases from organisms such as *Microbacterium* sp., *Bacillus* sp. RSJ-1, and *Acinetobacter* sp. with enzyme activity at 50 °C and alkaline pH (Lajis, 2018).

*Geobacillus* species are metabolically diverse extremophiles, and they exhibit a wide range of commercially useful extracellular enzymes, such as  $\alpha$ -amylases, xylanases, catalases, DNA polymerases, nucleases,  $\beta$ -galactosidases, hemicellulases, proteases, and lipases, principally. Recently, two bacterial strains that could produce lipases were isolated from an extreme environment, such as the geothermal waters of the "El Chichón" volcano's crater lake (Mexico). These strains were molecularly identified as belonging to the bacterial genus *Geobacillus* (Ovando-Chacon *et al.*, 2020). It is therefore necessary to optimize the production of these enzymes and to know their biochemical characteristics so that they can be used in specific biotechnological processes.

The objective of the present work was to perform an optimization of culture conditions of the strain *G. stearothermophilus* CHI1 and the biochemical characterization of the activity of its lipases based on temperature, pH, the presence of solvents, the presence of metal ions, and the affinity for substrate type.

## Materials and methods

### Microorganisms

The strain *G. stearothermophilus* CHI1 has been isolated, characterized, and molecularly identified as belonging to the bacterial genus *Geobacillus* (Ovando-Chacon *et al.*, 2020). The strain was maintained in Luria-Bertani modified medium (LBm) (Sodium Chloride, 0.5% m/v; peptone, 0.5% m/v; yeast extract, 0.3% m/v; pH 6.5) at 60 °C. The strain was designated as CHI1 and incorporated into ITTG culture collection.

### Lipase activity

Lipolytic activity was determined by the quantification of p-nitrophenol (pNP) released by enzymatic hydrolysis from the substrate p-nitrophenyl palmitate (pNPP). The pNPP production was determined from a calibration curve and the absorbance was determined at 402 nm. For this, the cell-free extract (supernatant of the bacterial growth medium centrifuged at 4000 rpm for 10 min) was used, and an emulsion was prepared from one volume of 0.79 mM pNPP dissolved in 2-propanol, with nine volumes of 0.1% w/v gum arabic dissolved in 50 mM phosphate buffer (pH 8). The enzymatic reaction was started by adding 300 µL of the cell-free extract to 2700 µL of the described emulsion and was incubated for 10 min at 60 °C. Finally, the absorbance was determined at 402 nm. A unit of enzyme activity (U) was defined as the amount of enzyme releasing 1 nmol of pNPP per minute under the assay conditions (Vorderwülbecke *et al.*, 1992). The pNPP esters were purchased from Sigma-Aldrich Mexico. All other reagents used in this work were of analytical grade.

### Optimization of physicochemical cultivation parameters

To maximize lipase production in *G. stearothermophilus* CHI1 cultures, an optimization design was established for the physicochemical parameters' temperature, pH, and agitation of the culture. The design employed was a central composite response surface design, with three variables and two levels for each variable. The design matrix consisted of 16 different treatments (eight factorial points, two axial points, and six central points), which were conducted in triplicate (Table 1).

Table 1. Treatments in the central composite experimental design for optimizing lipase production from *Geobacillus stearothermophilus* CHI1.

Treatment	Temperature (°C)	pH	Agitation (rpm)
1	50	3	50
2	70	3	50
3	50	5	50
4	70	5	50
5	50	3	150
6	70	3	150
7	50	5	150
8	70	5	150
9	43.2	4	100
10	76.8	4	100
11	60	2.3	100
12	60	5.7	100
13	60	4	16
14	60	4	184
16	60	4	100

Source: Autor's own elaboration.

The experimental unit was considered to be a culture of *G. stearothermophilus* CHI1 with 24 h of incubation under the conditions described. The response variable was lipolytic activity (U/mL), which was calculated as the average of three individual measurements for each treatment.

## Biochemical characterization of lipolytic activity

Biochemical characterization evaluated the effect of different temperature ranges, pH, presence of ions and solvents, and substrate specificity on enzyme activity. Extracellular lipolytic activity of cultures with 24 h of incubation was also determined. The results of the biochemical characterization were expressed in relative enzymatic activity, which was defined as the quotient of the enzymatic activity obtained in the presence of each of the factors evaluated concerning the enzymatic activity obtained in the control sample.

### Effect of temperature on lipolytic activity

Enzyme activity was determined by varying only the temperature of the enzyme assay, which ranged from 30 °C to 90 °C in 10 °C intervals. The results were expressed as relative enzyme activity (Ekinci *et al.*, 2016).

### Effect of pH on lipolytic activity

The enzyme activity was determined by varying the pH values, which ranged from pH 3 to pH 11 at intervals of 1. Different buffer solutions at a concentration of 50 mM were used for this assay: a citrate solution (pH 3-6), a phosphate solution (pH 7-8), and a carbonate solution (pH 9-11). These solutions replaced the 50 mM phosphate buffer solution of the enzymatic assay. The results were expressed as relative enzyme activity (Ekinci *et al.*, 2016).

### Effect of metallic ions on lipolytic activity

The effect of the metal ions Ba<sup>2+</sup>, Ca<sup>2+</sup>, Co<sup>2+</sup>, Cu<sup>2+</sup>, Fe<sup>3+</sup>, K<sup>+</sup>, Mn<sup>2+</sup>, and Na<sup>+</sup>, as well as the chelating agent EDTA on enzyme activity was determined. To do this, the cell-free extract was incubated at 37 °C for 15 min with each of the ions at a concentration of 10 mM; the enzyme assay was then performed for each condition. The results were expressed as relative enzyme activity (Ekinci *et al.*, 2016).

### Effect of solvents on the lipolytic activity

The effect of the acetone, chloroform, ethanol, isopropanol, and methanol on the lipolytic activity was determined. To do this, the cell-free extract was incubated at 37 °C for 15 min with each of the solvents at a final concentration of 30% (v/v), then the enzymatic assay was performed for each of the conditions. The results were expressed as relative enzyme activity (Ekinci *et al.*, 2016).

### Substrate specificity

The specificity towards pNPP esters of different acyl chain lengths (pNP octanoate, C8; pNP dodecanoate, C12; and pNP palmitate, C16) was evaluated using each of these substrates at a 0.79 mM concentration in the enzymatic assay methodology. The results were expressed as relative enzyme activity (Ekinci *et al.*, 2016).

## Statistical analysis

An analysis of variance was performed. A 95% confidence interval was used for the analysis of variance, and a *p*-value of less than 0.05 was considered statistically significant in the analyses. All tests were performed in triplicate. The response variable was the enzyme activity obtained as the average of three individual measurements of the treatments. Design-Expert version 11 and Statgraphics version 19 programs were used for data analysis and the generation of surface plots.

## Results

### Optimization of the lipase production

To maximize lipase production, a mathematical model must first be established to explain the relationship between the factors evaluated (temperature, pH, and agitation) and the response variable (enzyme activity) using a central composite experimental design. Four models were evaluated: a linear model, a linear model with two-factor interaction (2FI), a quadratic model, and a cubic model (Table 2).

Table 2. Models evaluated in the optimization of the physicochemical parameters of culture to maximize the lipase production of *Geobacillus stearothermophilus* CHI1.

Model	p-value of the model	R <sup>2</sup>	R <sup>2</sup> adjusted
Linear	0.4242	-0.0032	-0.2416
2FI	0.6975	-0.0418	-0.2549
Quadratic	< 0.0001	0.6563	0.5088
Cubic	0.1655	0.6824	0.4753

Source: Autor's own elaboration.

To choose the model with the best fit for the data, the following parameters were considered: the *p*-value of the model, the coefficient of determination R<sup>2</sup>, and the adjusted coefficient of determination R<sup>2</sup>, which quantifies the percentage of the variability present in the data explained by the model. In this case, the quadratic model is the model that best fits the data since it shows the statistical significance and the highest determination coefficient. The following is the equation resulting from the quadratic model (equation 1):

$$EA = -139.07 + 8.22 \cdot T - 67.30 \cdot \text{pH} + 0.67 \cdot A - 0.37 \cdot T \cdot \text{pH} + 0.004 \cdot T \cdot A - 0.06 \cdot \text{pH} \cdot A - 0.06 \cdot T^2 + 12.71 \cdot (\text{pH}^2 + 0.003 \cdot A^2) \quad (1)$$

where EA is the enzymatic activity, T is the temperature, A is the agitation, and pH of the culture medium.

The optimal conditions to maximize the enzymatic activity were 53.7 °C, pH of 5.7, and agitation of 80 rpm. The estimated response surface graph generated from the quadratic model shows the interaction between the pH and temperature factors (Figure 1). The graph shows two zones with higher enzymatic activity: the first corresponding to the interaction point between 65 °C and a pH of 2.3, where the enzymatic activity was approximately 45 U/mL; and the second corresponding to the interaction between 50 °C and a pH of 6.3, with the enzymatic activity of approximately 95 U/mL. In this work, in addition to the results obtained from the optimization of culture conditions, we found that the composition of the culture medium is not critical in the production of lipolytic enzymes in *G. stearothermophilus* CHI1 (data not shown).

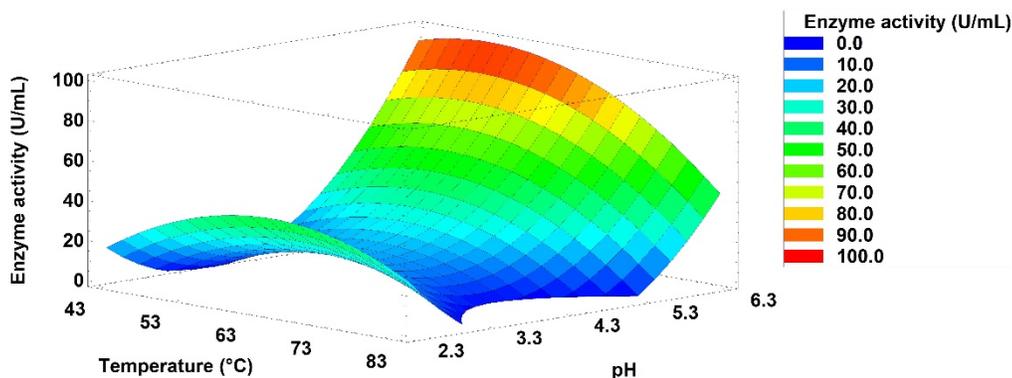


Figure 1. Estimated response surface plot of the *Geobacillus stearothermophilus* CHI1 extracellular enzyme activity versus the temperature and pH factors. Two zones with greater activity are shown: the first zone in the interaction of 65 °C and pH 2.3 (45 U/mL), and the second zone in the interaction between 50 °C and pH 6.3 (95 U/mL).  
Source: Autor's own elaboration.

The effect of different factors on the biochemical characteristics of the extracellular lipase activity present in *G. stearothermophilus* CHI1 were analyzed. Figure 2 shows the effect of temperature on relative enzyme activity for *G. stearothermophilus* CHI1. Higher relative lipolytic activity was observed at higher temperatures, with 2 points of maximum activity at 60 °C (84 U/mL) and 80 °C (85 U/mL). These results indicate the presence of thermophilic lipolytic enzymes in *G. stearothermophilus* CHI1. The enzymatic activity was evaluated.

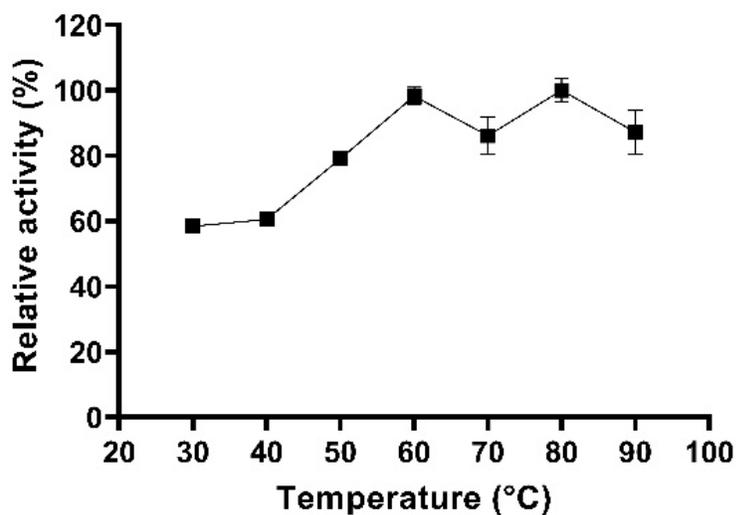


Figure 2. Effect of temperature on extracellular lipolytic activity *Geobacillus stearothermophilus* CHI1. The higher relative lipolytic activity was observed at 60 °C (84 U/mL) and 80 °C (85 U/mL).  
Source: Autor's own elaboration.

### Effect of pH on lipolytic activity

The effect of pH on the lipolytic activity of *G. stearothermophilus* CHI1 is shown in Figure 3, with higher lipolytic activity at pH values of 5 (310 U/mL), pH 9 (149 U/mL), and pH 11 (161 U/mL). These results indicate the presence of acid and alkaline lipases produced by *G. stearothermophilus* CHI1.

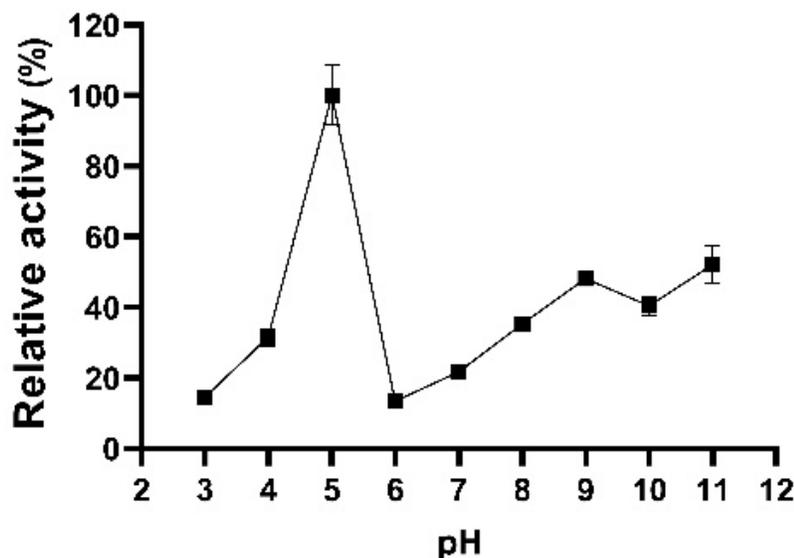


Figure 3. Effect of pH on the relative lipolytic activity of *Geobacillus stearothermophilus* CHI1. The higher relative lipolytic activity was observed at pH values of 5 (310 U/mL), pH 9 (149 U/mL), and pH 11 (161 U/mL).  
Source: Autor's own elaboration.

### Effect of metallic ions on lipolytic activity

The effect of different metal ions at a concentration of 10 mM on the relative lipolytic activity of *G. stearothermophilus* CHI1 is shown in Table 3. This table shows that there is a significant decrease in enzyme activity in the presence of  $\text{Na}^+$ ,  $\text{Mn}^{2+}$ , and  $\text{Fe}^{3+}$  ions, as well as the chelating agent EDTA and the control (absence of metallic ions). However, no improvement in enzymatic activity was present with the remaining ions:  $\text{Ca}^{2+}$ ,  $\text{K}^+$ ,  $\text{Co}^{2+}$ ,  $\text{Cu}^{2+}$ , and  $\text{Ba}^{2+}$ . The ions that most affected the activity were  $\text{Mn}^{2+}$  and  $\text{Fe}^{3+}$  ions with a 28% and 35% reduction in activity, respectively. These results suggest that the lipolytic enzymes produced by *G. stearothermophilus* CHI1 do not require metal cofactors for catalysis.

Table 3. Effect of metal ions on the relative lipolytic activity of *Geobacillus stearothermophilus* CHI1.

Treatment	Relative lipolytic activity (%)
Control	97.76 ± 2.05
$\text{Na}^+$	78.08 ± 4.36
$\text{K}^+$	92.39 ± 5.70
$\text{Ca}^{2+}$	87.25 ± 0.67
$\text{Mn}^{2+}$	70.02 ± 11.51
$\text{Fe}^{3+}$	63.76 ± 10.80
$\text{Co}^{2+}$	79.19 ± 0.68
$\text{Cu}^{2+}$	80.54 ± 1.78
$\text{Ba}^{2+}$	83.22 ± 11.47
EDTA	77.85 ± 3.08

Source: Autor's own elaboration.

## Effect of solvents on the lipolytic activity

The effect of the presence of solvents at a concentration of 30% (v/v) on the enzymatic activity of *G. stearothermophilus* CHI1 is shown in Table 4. Table 4 shows a decrease in lipolytic activity in the chloroform and methanol treatments, although it was not significant concerning the control. There was also a decrease in lipolytic activity in the presence of solvents, such as acetone, ethanol, and isopropanol of 40%, 25%, and 20%, respectively.

Table 4. Effect of different solvents on the lipolytic activity of *Geobacillus stearothermophilus* CHI1.

Solvents	Relative enzyme activity (%)
Control	97.49 ± 2.18
Acetone	57.86 ± 3.81
Chloroform	99.53 ± 2.36
Ethanol	73.27 ± 6.63
Isopropanol	77.36 ± 9.00
Methanol	87.74 ± 2.49

Source: Autor's own elaboration.

## Substrate specificity

Finally, the substrate affinity of the enzymes present in *G. stearothermophilus* CHI1 was evaluated using p-nitrophenol esters of different acyl chain lengths (Table 5). The highest enzymatic activity was obtained with pNP octanoate (C8), a medium-chain substrate with eight carbon atoms; the activity towards pNP dodecanoate (C12) and pNP palmitate (C16) represented 50% and 25% of the maximum activity obtained with the substrate pNP octanoate.

Table 5. Substrate affinity for the lipolytic enzymes from *Geobacillus stearothermophilus* CHI1.

Substrates*	Relative enzyme activity (%)
pNPO-C8	93.67 ± 5.50
pNPD-C12	45.80 ± 1.63
pNPP-C16	23.23 ± 0.49

\* pNPO-C8: p-nitrofenil-octanato; pNPD-C12: p-nitrofenil-dodecanato; p-nitrofenil hexadecanoato.  
Source: Autor's own elaboration.

The results obtained indicate the clear preference of the enzymes present in *G. stearothermophilus* CHI1 towards medium-chain substrates, which is characteristic of lipase enzymes and allows them to be differentiated from other enzyme groups such as esterases.

## Discussion

Lipases are hydrolytic enzymes that are in high demand in industrial processes, especially those enzymes from extremophilic organisms (Adrio & Demain, 2014). The *G. stearothermophilus* strain CHI1, isolated from a volcanic crater lake, produce lipolytic enzymes (Ovando-Chacón *et al.*, 2020). One of the objectives of the current work has been to optimize the culture conditions using statistical models to generate a higher production of these lipolytic enzymes. The mathematical model obtained allowed us to explain the relationship between the factors evaluated (temperature, pH, and agitation) and the response variable (enzyme activity), and to find the optimal conditions to maximize enzyme production. According to Figure 1 generated from the model, there are two zones with higher enzyme activity: one showing activity at a low pH and the other at high pH. If we also consider the results obtained in the biochemical characterization, specifically the effect of temperature and pH on enzyme activity (Figure 2 and 3), we see how this observation is presented again: concerning temperature, there are two points where the maximum enzyme activity occurs (at 60 °C and 80 °C), and for pH, something similar occurs at low pH (pH 3) and alkaline pH (pH 9-11). Taken together, the results of the optimization and characterization allow us to suggest the presence of at least two enzymes with lipolytic activity produced by the *G. stearothermophilus* strain CHI1, one of which shows higher catalysis at acid pH (acid lipase) and one more that shows higher catalysis at alkaline pH (alkaline lipase). In both cases catalysis at high temperatures is also shown, thus, considering both enzymes as thermophilic lipases.

There are few reports of biochemical characterization where more than one enzyme with lipolytic activity is presented. This is because most of these reports are made using different enzymatic purification methods, where the main objective is the isolation of a single protein (Dako *et al.*, 2012). During the purification process, there can be a loss of enzymatic activity and consequently a loss of protein material in each of the steps (e.g., with precipitation methods, an affinity to specific matrices, an exclusion by molecular size or charge, among others), which seriously hinders identifying more than one enzyme with lipolytic activity, as reported in the present work (Dako *et al.*, 2012). Moreover, two studies reported the presence of two different lipase enzymes produced by the *Bacillus subtilis* strain 168. These enzymes, named LipA and LipB, are regulated differently, have different inducers, and possess different biochemical characteristics (Eggert *et al.*, 2001, 2003). Within the genus *Geobacillus*, the presence of more than one enzyme with lipolytic activity has also been reported: a lipase enzyme and an esterase enzyme in *G. thermoleovorans*, with different structural and biochemical characteristics (Soliman *et al.*, 2007).

The results presented in this paper suggest that the enzymes with lipolytic activity produced by *G. stearothermophilus* CHI1 could also be regulated differently and could have different inducers. This is probably the main reason why we could not establish a fixed nutrient composition in the culture medium of *G. stearothermophilus* CHI1, even after using different statistical models.

The second objective of the present investigation was to biochemically characterize the lipolytic enzymes produced by *G. stearothermophilus* CHI1. Regarding the catalysis temperature, the lipolytic enzymes produced by *G. stearothermophilus* CHI1 showed activity at elevated temperatures (60 °C-80 °C). For the genus *Geobacillus*, lipolytic enzymes have been reported with an optimum catalysis temperature, ranging from 50 °C to 70 °C, depending on the organism: for the lipase enzymes of *G. stearothermophilus* strain 5 (Sifour *et al.*, 2010), *G. thermoleovorans* CCR11 (Castro-Ochoa *et al.*, 2005; Quintana-Castro *et al.*, 2009), *G. thermodenitrificans* AV-5 (Christopher *et al.*, 2015), and *G. thermoleovorans* YN (Soliman *et al.*, 2007), optimum temperatures between 60 °C and 65 °C were reported. For the lipases of *G. stearothermophilus* AH22 (Ekinici *et al.*, 2016) and *Geobacillus* sp. TW1 (Li & Zhang, 2005), optimum temperatures of 50 °C were reported. And for lipase from the *Geobacillus* sp. strain T1 (Leow *et al.*, 2007), an optimum catalysis temperature of 70 °C was reported. Regarding the pH of catalysis, the lipolytic enzymes produced by *G. stearothermophilus* CHI1 showed higher enzymatic activity at pH values of 5, pH 9, and pH 11. According to what has been reported in the literature for the genus *Geobacillus*, most of the studied lipases show an optimum pH of catalysis under alkaline conditions. For example, with lipase of *G. stearothermophilus* strain-5 (Berekaa *et al.*, 2009), the optimum pH was 7. Furthermore, the lipase enzymes of *G. stearothermophilus* AH22 (Ekinici *et al.*, 2016) and *Geobacillus* sp. TW1 (Li & Zhang, 2005), the optimum pH was 8. For the lipase enzymes of *G. thermoleovorans* CCR11 (Castro-Ochoa *et al.*, 2005; Quintana-Castro *et al.*, 2009), the *Geobacillus* sp. strain T1 (Leow *et al.*, 2007), and *G. thermodenitrificans* AV-5 (Christopher *et al.*, 2015), the optimum pH was 9. Finally, for the lipase enzyme of *G. thermoleovorans* YN (Soliman *et al.*, 2007), the optimum pH was 10. Since the vast majority of lipolytic enzymes present in *Geobacillus* showed activity exclusively under alkaline conditions, this study of *G. stearothermophilus* CHI1 may be the first report of these enzymes with enzymatic activity at acidic pH.

Considering now the effect of various metal ions on enzymatic activity, *G. stearothermophilus* CHI1 produces lipolytic enzymes that do not require metal ions as catalytic cofactors (Table 3). Most of the *Geobacillus* lipolytic enzymes described in the literature are calcium-dependent as they utilize these ions as cofactors for enzymatic activity, significantly enhancing catalysis (Ekinici *et al.*, 2016). Some examples of calcium-dependent lipolytic enzymes are the lipases of *G. thermoleovorans* CCR11 (Castro-Ochoa *et al.*, 2005), which increases its activity by 60% in the presence of calcium. The lipase present in *Geobacillus* sp. TW1 (Li & Zhang, 2005) increases its activity by 40%, and the esterase present in *Geobacillus thermoleovorans* YN (Soliman *et al.*, 2007) increases its activity by 50% in the presence of this ion. Some other divalent ions such as  $\text{Co}^{2+}$ ,  $\text{Cu}^{2+}$ , and  $\text{Ba}^{2+}$  have been shown to enhance enzyme activity in lipases such as those present in the *G. stearothermophilus* strain AH22 (Ekinici *et al.*, 2016). Since there was no significant improvement in catalytic activity in the presence of metal ions and there was little decrease of the enzymatic activity in the presence of the chelating agent EDTA, it is respectively suggested that the lipolytic enzymes present in *G. stearothermophilus* CHI1 may be metal ion-independent enzymes, also referred to as non-metalloenzymes.

Finally, regarding the effect of different organic solvents on enzymatic activity and regarding substrate specificity, the lipolytic enzymes produced by *G. stearothermophilus* CHI1 maintained more than 75% of their activity in the presence of chloroform, isopropanol, and methanol, and they showed greater preference towards medium-chain substrates (C8). Generally, the enzymes that are resistant to high temperatures are also resistant in the presence of organic solvents; this also applies to the lipolytic enzymes present in *G. stearothermophilus* CHI1, which, in addition to showing the highest activity at high temperatures, also maintains at least 60% of its enzymatic activity in the presence of the organic solvents evaluated. The stability of these enzymes in organic solvents (denoted by the presence of enzymatic activity in the conditions evaluated) suggests that they can be used in biocatalysis processes in the presence of nonaqueous solvents, such as in transesterification processes, in the synthesis of chiral compounds, and in the production of biofuels, among others (Lajis, 2018). The results of the specificity by substrate indicate a clear preference of the lipolytic enzymes produced by *G. stearothermophilus* CHI1 towards medium-chain substrates; these results agree with those reported for *G. thermoleovorans* CCR11 (Castro-Ochoa *et al.*, 2005), *G. thermoleovorans* YN (Soliman *et al.*, 2007), and *Geobacillus* sp. T1 (Leow *et al.*, 2007), with a higher affinity towards medium-chain substrates (C8, C10, and C12) compared with short-chain substrates (C6, C4, and C2).

According to the biochemical characteristics present in the lipolytic enzymes produced by *G. stearothermophilus* CHI1, they could be used in different biotechnological processes: 1) in the detergent industry, given their activity at alkaline pH; 2) in bioconversion reactions (fatty acid hydrolysis, esterification, alcoholysis, among others) and in the cosmetic industry, given their tolerance to organic solvents; 3) in the industrial generation of polyunsaturated fatty acids due to their high activity at elevated temperatures; 4) in the industrial production of biodiesel due to tolerance to organic solvents such as methanol; and 5) in wastewater treatment due to their activity at acidic pH (Salihu & Alam, 2015; Sharma & Kanwar, 2014).

## Conclusions

The enzymes evaluated in this work are characterized by having catalytic activity at elevated and acidic and alkaline pH, in the presence of solvents, independently of the presence of metal ions, and the enzymes have an affinity towards medium-chain substrates. The combination of the biochemical characteristics of these enzymes shows their potential for being used in different biotechnological processes that require high temperatures (e.g., the paper industry), the presence of solvents (e.g., biofuel production), and activity in alkaline conditions (e.g., detergent production). The results obtained in the optimization and characterization of the enzyme activity suggest the presence of more than one enzyme with lipolytic activity in *G. stearothermophilus* CHI1. Further studies are needed to fully elucidate the nature of the enzyme group (lipases or esterases) to which these proteins belong.

## Acknowledgments

CGH was supported by the Consejo Nacional de Ciencia y Tecnología de México. The authors thank the Consejo Nacional de Ciencia y Tecnología de México, the Tecnológico Nacional de México, and the Instituto de Ciencia Tecnología e Innovación del Estado de Chiapas for their financial support.

## Conflict of interest

The authors declare they have no financial interests.

## Author Contributions

Conceptualization, methodology, validation, formal analysis: C.O.G.H. and P.E.A.G.; software and data curation: M.A.A. and C.O.G.H.; investigation and methodology: C.O.G.H., M.A.A., L.M.C.V.C., and P.E.A.G.; resources: S.L.O.C.; writing—original draft preparation: C.O.G.H., M.A.A., L.M.C.V.C., and P.E.A.G.; supervision, project administration, funding acquisition: P.E.A.G. All authors have read and agreed to the published version of the manuscript.

## References

- Adetunji, A. I., & Olaniran, A. O. (2021). Production strategies and biotechnological relevance of microbial lipases: a review. *Brazilian Journal Microbiology*, *52*, 1257–1269. <https://doi.org/10.1007/s42770-021-00503-5>
- Adrio, J. L., & Demain, A. L. (2014). Microbial enzymes: tools for biotechnological processes. *Biomolecules*, *4*(1), 117–139. <https://doi.org/10.3390/biom4010117>
- Barik, A., Sen, S. K., Rajhans, G., & Raut, S. (2022). Purification and optimization of extracellular lipase from a novel strain *Kocuria flava* Y4. *International Journal of Analytical Chemistry*, *2022*, 1–10. <https://doi.org/10.1155/2022/6403090>
- Behera, A. R., Veluppall, A., & Dutta, K. (2019). Optimization of physical parameters for enhanced production of lipase from *Staphylococcus hominins* using response surface methodology. *Environmental Science and Pollution Research*, *26*, 34277–34284. <https://doi.org/10.1007/s11356-019-04304-0>
- Berekaa, M. M., Zaghoul, T. I., Abdel-Fattah, Y. R., Saeed, H. M., & Sifour, M. (2009). Production of a novel glycerol-inducible lipase from thermophilic *Geobacillus stearothermophilus* strain-5. *World Journal of Microbiology and Biotechnology*, *25*, 287–294. <https://doi.org/10.1007/s11274-008-9891-3>
- Castro-Ochoa, L. D., Rodríguez-Gómez, C., Valerio-Alfaro, G., & Oliart, R. (2005). Screening, purification, and characterization of the thermoalkalophilic lipase produced by *Bacillus thermoleovorans* CCR11. *Enzyme and Microbial Technology*, *37*, 648–654. <https://doi.org/10.1016/j.enzmictec.2005.06.003>
- Christopher, L. P., Zambare, V. P., Zambare, A., Kumar, H., & Malek, L. (2015). A thermo-alkaline lipase from a new thermophile *Geobacillus thermodenitrificans* AV-5 with potential application in biodiesel production. *Journal of Chemical Technology & Biotechnology*, *90*, 2007–2016. <https://doi.org/10.1002/jctb.4678>
- Dako, E., Bernier, A. M., Dadie, A. T., & Jankowski, C. K. (2012). The problems associated with enzyme purification. In D. Ekinici (ed.), *Chemical Biology* (pp. 19–40). Intech. <https://doi.org/10.5772/33307>
- Eggert, T., van Pouderoyen, G., Dijkstra, B. W., & Jaeger, K. E. (2001). Lipolytic enzymes LipA and LipB from *Bacillus subtilis* differ in regulation of gene expression, biochemical properties, and three-dimensional structure. *FEBS Letters*, *502*, 89–92. [https://doi.org/10.1016/S0014-5793\(01\)02665-5](https://doi.org/10.1016/S0014-5793(01)02665-5)
- Eggert, T., Brockmeier, U., Droge, M. J., Quax, W. J., & Jaeger, K. E. (2003). Extracellular lipases from *Bacillus subtilis*: regulation of gene expression and enzyme activity by amino acid supply and external pH. *FEMS Microbiology Letters*, *225*, 319–324. [https://doi.org/10.1016/S0378-1097\(03\)00536-6](https://doi.org/10.1016/S0378-1097(03)00536-6)
- Ekinici, A. P., Dinçer, B., Baltaş, N., & Adıgüzel, A. (2016). Partial purification and characterization of lipase from *Geobacillus stearothermophilus* AH22. *Journal of Enzyme Inhibition and Medical Chemistry*, *31*, 325–331. <https://doi.org/10.3109/14756366.2015.1024677>
- Gamboa-Melendez, H., Larroude, M., Park, Y. K., Trebul, P., Nicaud, J. M., & Ledesma-Amaro, R. (2018). Synthetic biology to improve the production of lipases and esterases (Review). *Methods in Molecular Biology*, *1835*, 229–242. [https://doi.org/10.1007/978-1-4939-8672-9\\_13](https://doi.org/10.1007/978-1-4939-8672-9_13)
- Godoy, C. A., Pardo-Tamayo, J. S., & Barbosa, O. (2022). Microbial lipases and their potential in the production of pharmaceutical building blocks. *International Journal of Molecular Science*, *23*, 9933. <https://doi.org/10.3390/ijms23179933>
- Lajis, A. F. B. (2018). Realm of thermoalkaline lipases in bioprocess commodities. *Journal of Lipids*, *2018*, 1–22. <https://doi.org/10.1155/2018/5659683>
- Leow, T. C., Rahman, R. N. Z. R. A., Basri, M., & Salleh, A. B. (2007). A thermoalkaliphilic lipase of *Geobacillus* sp. T1. *Extremophiles*, *11*, 527–535. <https://doi.org/10.1007/s00792-007-0069-y>

- Li, H., & Zhang, X. (2005). Characterization of thermostable lipase from thermophilic *Geobacillus* sp. TW1. *Protein Expression and Purification*, 42(1), 153–159. <https://doi.org/10.1016/j.pep.2005.03.011>
- Lim, S. Y., Steiner, J. M., & Cridge, H. (2022). Lipases: It's not just pancreatic lipase!. *American Journal of Veterinary Research*, 83(8), 1-8. <https://doi.org/10.2460/ajvr.22.03.0048>
- Mahfoudhi, A., Benmabrouk, S., Fendri, A., & Sayari, A. (2022). Fungal lipases as biocatalysts: a promising platform in several industrial biotechnology applications. *Biotechnology and Bioengineering*, 119(12), 3370-3392. <https://doi.org/10.1002/bit.28245>
- Oliart-Ros, R. M., Badillo-Zeferino, G. L., Quintana-Castro, R., Ruíz-López, I. I., Alexander-Aguilera, A., Domínguez-Chávez, J. G., Khan, A. A., Nguyen, D. D., Nadda, A. K., & Sánchez-Otero, M. G. (2021). Production and characterization of cross-linked aggregates of *Geobacillus thermoleovorans* CCR11 thermoalkaliphilic recombinant lipase. *Molecules*, 26(24), 7569. <https://doi.org/10.3390/molecules26247569>
- Ovando-Chacon, S. L., Tacias-Pascacio, V. G., Ovando-Chacon, G. E., Rosales-Quintero, A., Rodriguez-Leon, A., Ruiz-Valdiviezo, V. M., & Servin-Martinez, A. (2020). Characterization of thermophilic microorganisms in the geothermal water flow of El Chichón volcano crater lake. *Water*, 12, 2172. <https://doi.org/10.3390/w12082172>
- Pohanka, M. (2019). Biosensors and bioassays based on lipases, principles and applications, a review. *Molecules*, 24(3), 616. <https://doi.org/10.3390/molecules24030616>
- Quintana-Castro, R., Díaz, P., Valerio-Alfaro, G., García, H. S., & Oliart-Ros, R. (2009). Gene cloning, expression, and characterization of the *Geobacillus thermoleovorans* CCR11 thermoalkaliphilic lipase. *Molecular Biotechnology*, 42, 75–83. <https://doi.org/10.1007/s12033-008-9136-6>
- Rmili, F., Hadrich, B., Chamkha, M., Sayari, A., & Fendri, A. (2022). Optimization of organic solvent-tolerant lipase production by *Staphylococcus capitis* SH6. Immobilization for biodiesel production and biodegradation of waste greases. *Preparative Biochemistry & Biotechnology*, 52, 108–122. <https://doi.org/10.1080/10826068.2021.1920034>
- Salihi, A., & Alam, Z. (2015). Solvent tolerant lipases: a review. *Process Biochemistry*, 50(1), 86–96. <https://doi.org/10.1016/j.procbio.2014.10.019>
- Sharma, S., & Kanwar, S. S. (2014). Organic solvent tolerant lipases and applications. *The Scientific World Journal*, 2014, 625258. <https://doi.org/10.1155/2014/625258>
- Sifour, M., Saeed, H. M., Zaghoul, T. I., Berekaa, M. M., & Abdel-Fatt, Y. R. (2010). Purification and properties of a lipase from thermophilic *Geobacillus stearothermophilus* Strain-5. *International Journal of Biological Chemistry*, 4(4), 203–212. <https://doi.org/10.3923/ijbc.2010.203.212>
- Soliman, N. A., Knoll, M., Abdel-Fattah, Y. R., Schmid, R. D., & Lange, S. (2007). Molecular cloning and characterization of thermostable esterase and lipase from *Geobacillus thermoleovorans* YN isolated from desert soil in Egypt. *Process Biochem*, 42(7), 1090–1100. <https://doi.org/10.1016/j.procbio.2007.05.005>
- Vivek, K., Sandhia, G. S., & Subramaniyan, S. (2022). Extremophilic lipases for industrial applications: a general review. *Biotechnology Advances*, 60, 108002. <https://doi.org/10.1016/j.biotechadv.2022.108002>
- Vorderwülbecke, T., Kieslich, K., & Erdmann, H. (1992). Comparison of lipases by different assays. *Enzyme and Microbial Technology*, 14(8), 631–639. [https://doi.org/10.1016/0141-0229\(92\)90038-P](https://doi.org/10.1016/0141-0229(92)90038-P)